



General Comparison of Compressed Air, Supercritical CO₂, and Steam at 100 C in Terms of Density

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Below is a general comparison of the densities of compressed air, supercritical CO2, and steam at 100°C to understand how they might behave when expanding through a turbine:

1. Density Comparison
- Compressed Air:

• The density of air at standard temperature and pressure (STP: 1 atm, 25°C) is approximately 1.2 kg/m³.

• When air is compressed, its density increases based on the pressure applied. For instance, at 10 atm and room temperature, the density can be roughly 12 kg/m³. However, it is still significantly less dense than both supercritical CO2 and liquid water.
- Supercritical CO2:

• Supercritical CO2 (sCO2) is CO2 above its critical point (31.1°C and 1071 psi). In its supercritical state, it exhibits properties between those of a gas and a liquid.

• The density of supercritical CO2 typically ranges from 100 to 600 kg/m³, depending on temperature and pressure. For example, at 30 MPa (about 4350 psi) and 40°C, sCO2 can have a density of around 200-500 kg/m³.

• Compared to steam at 100°C, sCO2 is roughly 10 to 100 times more dense, depending on the exact conditions. This higher density enables more efficient momentum transfer to turbine blades, enhancing energy extraction.
- Steam at 100°C:

• The density of steam at 100°C (1 atm) is approximately 0.6 kg/m³. When heated and pressurized, the density of steam increases, but under typical steam turbine operating conditions, it is still less dense compared to supercritical CO2.

• Steam can have a density of 30-60 kg/m³ when pressurized to levels used in power plants. However, it still remains far less dense than supercritical CO2 in most power generation scenarios.
2. Key Comparisons and Impact on Turbine Behavior
- Density and Energy Extraction:

• Supercritical CO2: The high density of sCO2 allows for effective energy extraction with compact turbine designs. It provides higher specific work output and reduced compressor work relative to its expansion in a turbine.

• Steam: Due to its lower density, steam requires significantly larger turbine volumes to achieve similar energy output as sCO2. The rapid expansion of steam (with large changes in specific volume) is beneficial for generating work, but the physical size of the components tends to be larger.

• Compressed Air: The density of air can be increased by compression, but it will still remain lower than sCO2 at the same pressures. Lower density means lower specific work output and higher turbine speeds or larger turbine components are needed to efficiently extract energy.
- Application Context:

• Supercritical CO2 is favored in compact power generation systems and waste heat recovery because its high density allows for compact and highly efficient turbines. It is used in advanced power cycles to achieve high efficiency.

• Steam is typically used in traditional power generation like Rankine cycles because it can take advantage of latent heat during phase changes, making it efficient for large-scale power generation with significant heat input.

• Compressed Air is used in open Brayton cycles like those in gas turbines. While air can be pressurized to improve energy density, its inherently lower density means less efficiency for a given turbine volume compared to sCO2.
- Summary
- Supercritical CO2: Roughly 10 to 100 times more dense than steam at 100°C, depending on pressure and temperature. The high density allows for efficient power conversion, making it suitable for

Expansion Through a Turbine

The behavior of a working fluid (such as CO₂ gas, steam, or air) expanding through a turbine depends on several properties, with density being one of the critical factors. Density affects how efficiently energy is converted as the fluid expands, impacting the design and operation of the turbine. Here is a comparison of CO₂ gas, steam, and air in terms of their relative densities and how these differences influence their expansion through a turbine.

1. Relative Densities

- Carbon Dioxide (CO₂) Gas:

- At standard conditions (1 atm and 25°C), the density of CO₂ gas is approximately 1.98 kg/m³. When supercritical, CO₂ has much higher densities (ranging from 100 to 600 kg/m³, depending on temperature and pressure).
- In its supercritical state, CO₂ is dense and behaves like a fluid, making it an efficient medium for power cycles due to the high energy extraction potential. Its relatively high density allows it to effectively impart momentum to turbine blades.

- Steam (Water Vapor):

- The density of steam at atmospheric pressure and 100°C is approximately 0.6 kg/m³. The density changes significantly depending on temperature and pressure. In high-pressure turbines, steam can be much denser, around 30–60 kg/m³.
- Steam density changes substantially as it expands in a turbine, which can lead to rapid volumetric changes. This is beneficial in power cycles like the Rankine cycle, where high energy can be extracted from the phase changes (liquid to vapor).

- Air:

- At standard conditions (1 atm and 25°C), air has a density of 1.2 kg/m³. Compared to CO₂, air is less dense, but it is also a common working fluid for various cycles due to its availability.
- Air's density is lower than that of CO₂ and steam, making it less effective in transferring energy per unit volume in a turbine. This difference affects how the turbine extracts work from air, often requiring larger turbine stages to achieve efficient expansion.

2. Expansion Through a Turbine

- Behavior of CO₂ Gas in a Turbine:

- Supercritical CO₂ (sCO₂) has unique properties that make it highly efficient for energy extraction. Its density is higher than air and steam, especially in the supercritical region, which allows for smaller turbines and more compact systems.
- sCO₂ behaves more like a dense fluid than a gas, resulting in efficient heat transfer and better use of pressure drops in a turbine. This makes it particularly attractive for applications in power generation systems where space and efficiency are critical, such as in waste heat recovery and nuclear power.

- Behavior of Steam in a Turbine:

- Steam expands significantly as it moves through a turbine due to its change in phase and density. This expansion causes a large increase in specific volume, which provides a significant amount of work.
- Steam turbines are well established in power generation, largely due to their ability to utilize the energy from phase changes efficiently. The steam's lower density compared to sCO₂ requires larger turbine stages to accommodate the rapid volumetric expansion, which is why steam turbines tend to be larger in size.

- Behavior of Air in a Turbine:

- Air, with its lower density, results in lower energy extraction per unit volume. To compensate for this, air turbines (such as those used in gas turbine engines) operate at much higher pressures and temperatures.

Expansion Through a Turbine

The behavior of a working fluid (such as CO2 gas, steam, or air) expanding through a turbine depends on several properties, with density being one of the critical factors. Density affects how efficiently energy is converted as the fluid expands, impacting the design and operation of the turbine. Here is a comparison of CO2 gas, steam, and air in terms of their relative densities and how these differences influence their expansion through a turbine.

1. Relative Densities
- Carbon Dioxide (CO2) Gas:

• At standard conditions (1 atm and 25°C), the density of CO2 gas is approximately 1.98 kg/m³. When supercritical, CO2 has much higher densities (ranging from 100 to 600 kg/m³, depending on temperature and pressure).

• In its supercritical state, CO2 is dense and behaves like a fluid, making it an efficient medium for power cycles due to the high energy extraction potential. Its relatively high density allows it to effectively impart momentum to turbine blades.
- Steam (Water Vapor):

• The density of steam at atmospheric pressure and 100°C is approximately 0.6 kg/m³. The density changes significantly depending on temperature and pressure. In high-pressure turbines, steam can be much denser, around 30–60 kg/m³.

• Steam density changes substantially as it expands in a turbine, which can lead to rapid volumetric changes. This is beneficial in power cycles like the Rankine cycle, where high energy can be extracted from the phase changes (liquid to vapor).
- Air:

• At standard conditions (1 atm and 25°C), air has a density of 1.2 kg/m³. Compared to CO2, air is less dense, but it is also a common working fluid for various cycles due to its availability.

• Air's density is lower than that of CO2 and steam, making it less effective in transferring energy per unit volume in a turbine. This difference affects how the turbine extracts work from air, often requiring larger turbine stages to achieve efficient expansion.

2. Expansion Through a Turbine
- Behavior of CO2 Gas in a Turbine:

• Supercritical CO2 (sCO2) has unique properties that make it highly efficient for energy extraction. Its density is higher than air and steam, especially in the supercritical region, which allows for smaller turbines and more compact systems.

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• Steam expands significantly as it moves through a turbine due to its change in phase and density. This expansion causes a large increase in specific volume, which provides a significant amount of work.

• Steam turbines are well established in power generation, largely due to their ability to utilize the energy from phase changes efficiently. The steam's lower density compared to sCO2 requires larger turbine stages to accommodate the rapid volumetric expansion, which is why steam turbines tend to be larger in size.
- Behavior of Air in a Turbine:

• Air, with its lower density, results in lower energy extraction per unit volume. To compensate for this, air turbines (such as those used in gas turbine engines) operate at much higher pressures and temperatures.

• The lower density of air means that for a given mass flow rate, larger turbines or higher rotational speeds are required to achieve efficient energy conversion compared to denser working fluids like sCO2.



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General Comparison of Compressed Air, Supercritical CO2, and Steam at 100°C Based on Power or Energy Density

Power density and energy density are important metrics that indicate how much energy a working fluid can deliver during its expansion in a turbine. Here, we compare compressed air, supercritical CO2, and steam at 100°C to evaluate their effectiveness in energy extraction and power generation:

1. Energy Density and Power Density Overview

- **Energy Density:** The amount of energy per unit mass or volume that a working fluid can provide when expanding through a turbine.
- **Power Density:** The rate at which energy can be delivered by the fluid per unit volume or mass flow rate during turbine operation.

2. Energy Density Comparison

Supercritical CO2 (sCO2)

- **Energy Density:** Supercritical CO2 has a high energy density due to its high density and favorable thermodynamic properties at supercritical conditions (above 31.1°C and 1071 psi). At supercritical pressures and moderate temperatures, sCO2 exhibits characteristics similar to both a gas and a liquid, which results in more efficient energy transfer and higher energy density compared to steam and air.
- The energy density of sCO2 in a closed-loop cycle is higher due to reduced compression work, which allows more energy to be extracted during turbine expansion.

Steam at 100°C

- **Energy Density:** The energy density of steam at 100°C and atmospheric pressure is lower than that of sCO2, as steam at this temperature has a relatively low density (~0.6 kg/m³). In a typical Rankine cycle, steam must be pressurized to increase its energy density, which also adds complexity and energy requirements for the boiler and pump.
- Steam, especially in high-pressure applications, can still provide significant energy, but it is less efficient in energy per unit volume compared to sCO2 due to the need for phase change and latent heat utilization.

Compressed Air

- **Energy Density:** Compressed air has a lower energy density compared to both sCO2 and steam. This is mainly due to its relatively low density, even when compressed to higher pressures (e.g., 10 -20 atm). The energy required to compress air to high pressures significantly reduces the overall energy efficiency and density compared to sCO2.
- The specific energy of compressed air systems is typically lower than that of sCO2 and steam, which means that more air volume is needed to produce the same amount of power as sCO2 or high-pressure steam.

3. Power Density Comparison

Supercritical CO2 (sCO2)

- **Power Density:** sCO2 has a very high power density, allowing for compact and efficient turbine systems. Due to its high density and favorable thermodynamic characteristics, sCO2 can produce more power per unit volume of working fluid. This is especially useful in power generation systems where space and weight are constrained, such as in waste heat recovery and advanced power cycles.
- The high enthalpy drop in sCO2 as it expands through the turbine leads to a high power output, which also improves the overall power density.

Steam at 100°C

- **Power Density:** The power density of steam depends on its pressure and temperature. At 100°C and atmospheric pressure, the power density is relatively low due to the low density of steam. In power generation applications, steam is typically pressurized to significantly increase its power density, making it a suitable working fluid for large-scale power plants.
- The expansion ratio of steam leads to large turbine volumes, as it undergoes a significant increase in specific volume during phase change. This results in a lower power density compared to sCO2.



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