



Thermal Battery Power Output Using ORC or Supercritical CO₂ Turbine Generator at 40,000 BTU/kWh

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<https://infinityturbine.com/infinity-turbine-sand-salt-battery-build.html>

Explore how a 40-foot container thermal battery filled with sand and salt can generate electricity using an Organic Rankine Cycle or supercritical CO₂ turbine with a heat rate of 40,000 BTU per kilowatt-hour. Includes power output estimates for 1-hour, 12-hour, and 24-hour discharge cycles.



This webpage QR code

PDF Version of the webpage (maximum 10 pages)

Thermal Battery Power Output Using ORC or Supercritical CO₂ Turbine Generator at 40,000 BTU/kWh

By converting thermal energy stored in a sand-salt thermal battery through an ORC or supercritical CO₂ turbine, a standard 40-foot container can produce nearly 1,250 kWh of electricity. Learn how this system performs over short and long discharge durations using a 40,000 BTU/kWh heat rate.

Thermal Battery Electricity Output Using ORC or Supercritical CO₂ Turbines

A 40-foot shipping container filled with a 50/50 mix of sand and salt can serve as a high-capacity thermal battery. By integrating this system with a turbine generator—either an Organic Rankine Cycle (ORC) or a supercritical CO₂ expander—the stored heat can be converted into usable electricity.

In this analysis, we calculate the expected power generation based on a standard heat rate of 40,000 BTU per kilowatt-hour (kWh), which is common for low- to medium-efficiency waste heat recovery turbines.

Stored Thermal Energy

The container holds approximately 16.95 million BTU of usable thermal energy, based on the thermal mass and a temperature swing from 100°F to 500°F:

BTU: 16,947,200

Metric equivalent: ≈ 4,965 kWh of heat energy

Heat Rate and Conversion to Electricity

Using the heat rate of 40,000 BTU/kWh, the maximum potential electric output is:

$> 16,947,200 \text{ BTU} \div 40,000 \text{ BTU/kWh} = 423.68 \text{ kWh}$

This is the total electrical energy that can be extracted if the turbine generator maintains a 40,000 BTU/kWh heat rate.

Output by Timeframe

To understand how this energy can be used over time, we calculate the constant power output for 1-hour, 12-hour, and 24-hour periods.

1-Hour Discharge



Thermal Battery as Hydraulic Energy Source

Assessing your concept of using a thermal battery (sand-salt mixture in a 40-foot container) to heat CO₂ to supercritical conditions and then drive a hydraulic pump, we can evaluate:

1. Available thermal energy
2. CO₂ heating and expansion capacity
3. Mechanical output in terms of hydraulic horsepower (HP) and gallons per minute (GPM)
4. Feasibility for 1-hour, 12-hour, and 24-hour continuous discharge

1. Thermal Energy Storage Recap

From your optimized container setup:

Stored thermal energy:

≈ 16.95 million BTU

≈ 4,965 kWh

≈ 17.9 billion joules (MJ: 17,900)

2. CO₂ Expansion to Drive Hydraulic Pump

Supercritical CO₂ Key Properties:

Critical point: ~31°C (88°F) and 1,070 psi

Operating target: ~500°F (260°C), ~2,000–3,000 psi

Expansion drives turbine or piston, which powers a hydraulic pump

Let's assume:

CO₂ loop runs through a high-temperature heat exchanger inside the thermal battery

Supercritical CO₂ drives an expander (piston or turbine)

Expander is mechanically coupled to a hydraulic pump

3. Efficiency Assumptions

We'll break the energy flow into conversion steps:

Step	Process	Efficiency Estimate
A	Heat → CO ₂ mechanical energy	20–25%
B	Mechanical → hydraulic energy	90%
Overall	Thermal → hydraulic output	~20% x 0.9 = 18%

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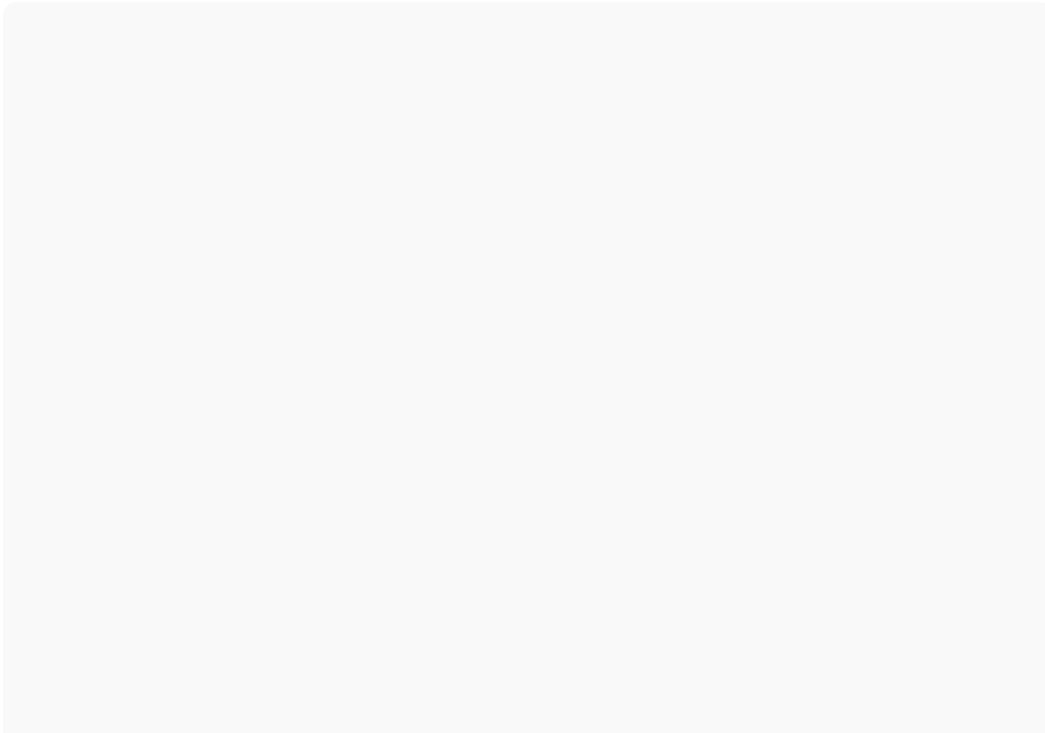
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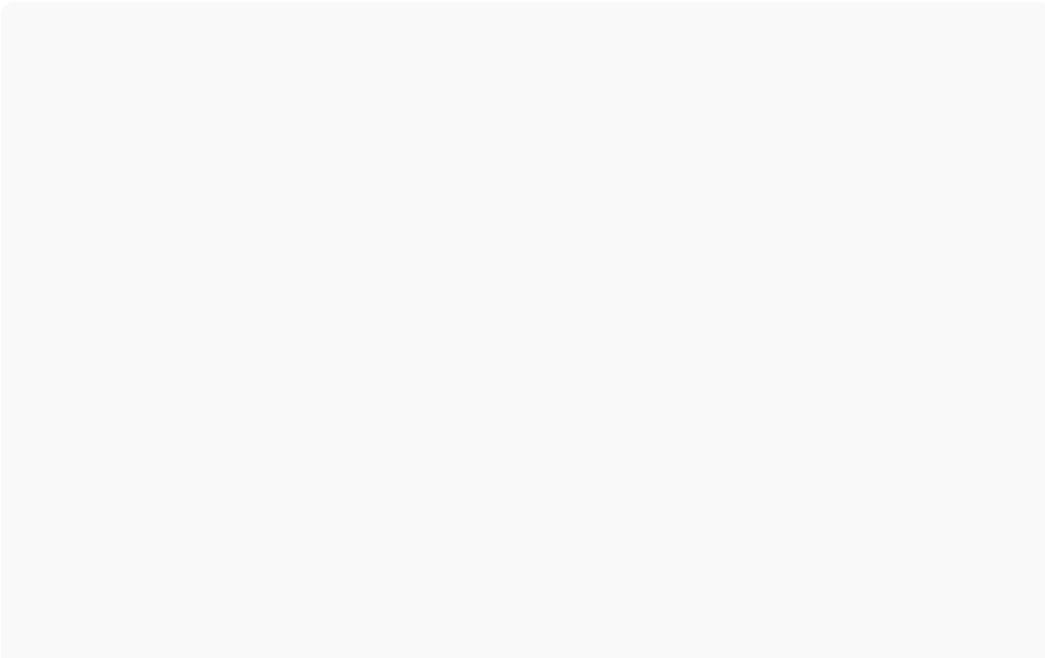
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Enhanced Power Output from Thermal Battery Using ORC or Supercritical CO₂ Turbine at 20,000 BTU/kWh

Page Title:

Enhanced Power Output from Thermal Battery Using ORC or Supercritical CO₂ Turbine at 20,000 BTU/kWh

Meta Description:

Assess how a sand-salt thermal battery inside a 40-foot shipping container can generate over 840 kWh of electricity using an ORC or supercritical CO₂ turbine with a heat rate of 20,000 BTU per kilowatt-hour. Includes power delivery profiles for 1-hour, 12-hour, and 24-hour discharge cycles.

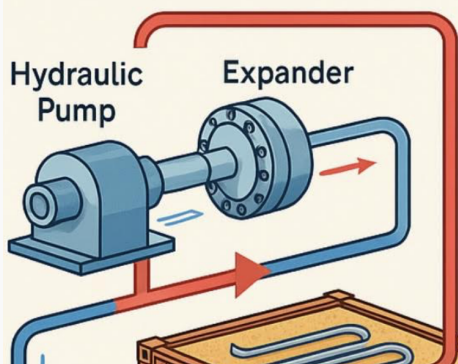
Teaser:

Using a turbine generator with a heat rate of 20,000 BTU per kWh, the thermal battery in a 40-foot shipping container can produce more than 840 kWh of electricity. Discover how this high-efficiency conversion method affects power output over different operating durations.

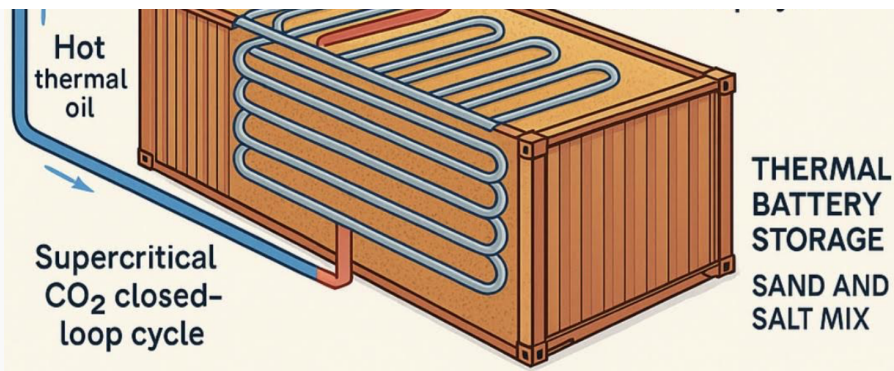
High-Efficiency Thermal Battery Power Generation Using 20,000 BTU/kWh Heat Rate

A 40-foot shipping container filled with a 50/50 mixture of sand and salt functions as a high-density thermal battery. By coupling this system with a high-efficiency Organic Rankine Cycle (ORC) or supercritical CO₂ turbine, stored thermal energy can be converted into electrical energy. In this article, we evaluate the power output using a heat rate of 20,000 BTU per kilowatt-hour, reflecting an efficient turbine design.

THERMAL BATTERY FOR PRODUCING HYDRAULIC POWER



- Container filled with 50/50 mix of sand and salt
- Thermal oil circulated through steel pipes to heat sand-salt mix
- Heated medium transfers heat to supercritical CO₂ in closed-loop cycle



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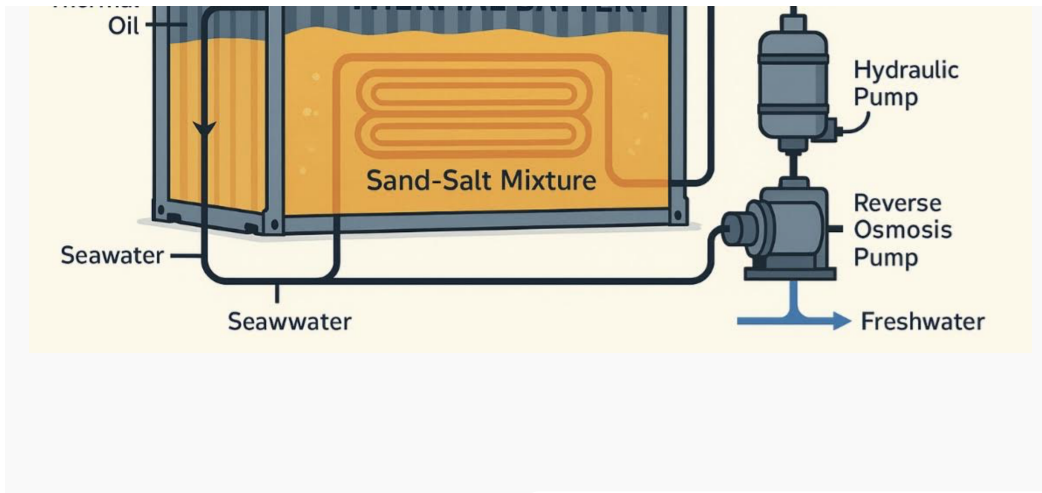
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Solar-Powered Thermal Battery Drives Reverse Osmosis Freshwater Generation





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Solar Thermal Battery Drives High-Capacity Chilling for Off-Grid Cooling

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Solar Thermal Battery Drives High-Capacity Chilling for Off-Grid Cooling

Meta Description:

Discover how a 40-foot sand-salt thermal battery with concentrated solar input can power thermally or electrically driven chillers, delivering over 1,500 tons of cooling-hours—ideal for off-grid or industrial cooling.

Teaser:

A single thermal battery using sand-salt storage and rooftop solar collectors can provide over 1,500 tons of cooling, delivering up to 127 tons per hour using high-efficiency chillers. This scalable system offers renewable-driven cooling for industrial or remote applications.

Solar Thermal Battery as a Renewable Cooling Source

A 40-foot shipping container filled with a 50/50 sand and salt mixture, equipped with concentrated solar panels and internal thermal oil piping, becomes a robust off-grid thermal battery. When paired with a chiller system—either thermally or electrically driven—it delivers substantial and sustained cooling capacity.

Two Approaches to Cooling

Absorption Chiller (Thermal-Driven)

Uses thermal energy directly via lithium bromide absorption system

COP: ~0.7

Cooling output: ~989 tons of cooling-hours

Ideal for locations where electricity is limited

Electric Chiller (Power-Driven)

Uses power generated from heated supercritical CO₂ driving a turbine

COP: ~6.0

Cooling output: ~1,525 tons of cooling-hours

Delivers up to 127 tons/hour over 12 hours

Cooling Delivery Profiles

System Type 1-Hour Burst 12-Hour Avg 24-Hour Avg

Absorption Chiller 989 tons 82.4 tons/hr 41.2 tons/hr

Electric Chiller 1,525 tons 127.1 tons/hr 63.5 tons/hr

Use Cases

Data center cooling

Cold storage in agriculture

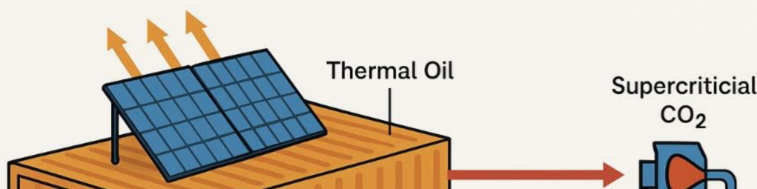
Emergency or disaster zone chilling

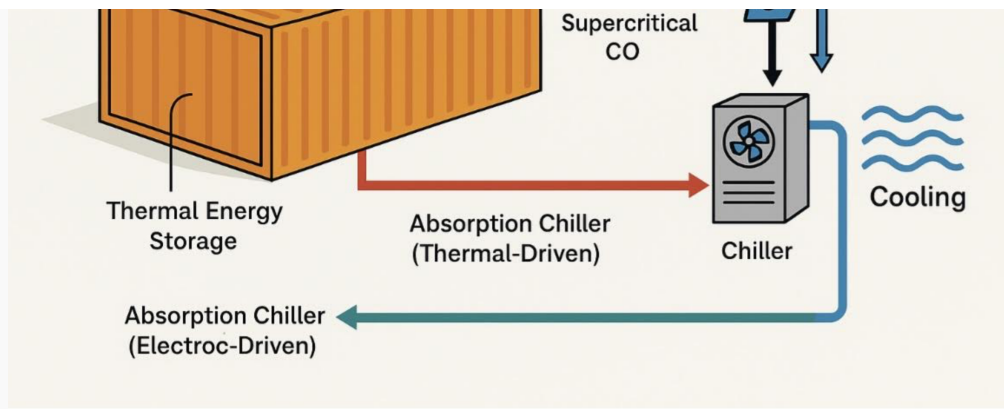
Industrial HVAC in off-grid regions

Conclusion

This is a conceptual diagram of a solar thermal battery system. It shows a solar collector (blue panels) on a wooden structure, labeled 'Thermal Oil'. An arrow points from the solar collector to a turbine labeled 'Supercritical CO₂'.

Solar Thermal Battery Cooling





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To compare the thermal energy density of sand, salt, and a 50/50 sand-salt mixture, we evaluate:

1. Density (mass per unit volume)
2. Specific heat capacity (energy per unit mass per °C or °F)
3. Thermal energy density (energy stored per unit volume per °C or °F)

1. Material Properties

Property	Sand	Salt (NaCl)	50/50 Mix
Bulk Density (kg/m ³)	~1,600	~1,200	~1,400
Specific Heat (kJ/kg·K)	~0.80	~0.85	~0.825 (average)
Specific Heat (BTU/lb·°F)	~0.19	~0.22	~0.205

2. Thermal Energy Density (per m³)

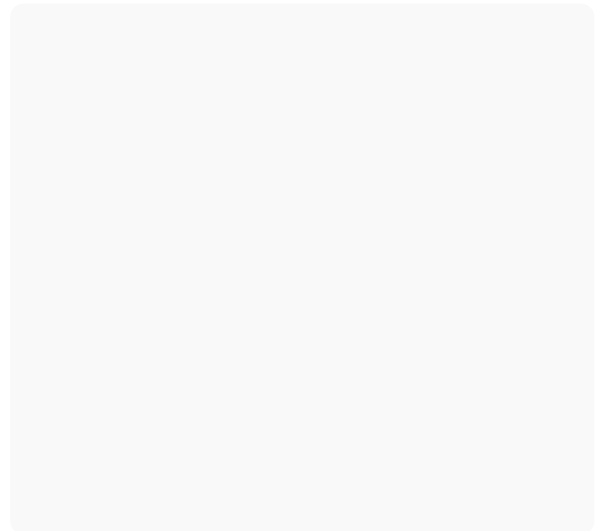
> Thermal Energy Density = Density × Specific Heat

Metric (kJ/m³·K):

Sand: $1,600 \times 0.80 = 1,280 \text{ kJ/m}^3\cdot\text{K}$
 Salt: $1,200 \times 0.85 = 1,020 \text{ kJ/m}^3\cdot\text{K}$
 50/50 Mix: $1,400 \times 0.825 = 1,155 \text{ kJ/m}^3\cdot\text{K}$

Imperial (BTU/ft³·°F):

> Convert using:



$$> 1 \text{ kg/m}^3 \times 0.00006243 = \text{lb/ft}^3$$

$$> 1 \text{ kJ} = 0.9478 \text{ BTU}$$

Sand:

$$\sim 100 \text{ lb/ft}^3 \times 0.19 = 19 \text{ BTU/ft}^3 \cdot ^\circ\text{F}$$

Salt:

$$\sim 75 \text{ lb/ft}^3 \times 0.22 = 16.5 \text{ BTU/ft}^3 \cdot ^\circ\text{F}$$

50/50 Mix:

$$\sim 87.5 \text{ lb/ft}^3 \times 0.205 = 17.94 \text{ BTU/ft}^3 \cdot ^\circ\text{F}$$

Summary: Thermal Energy Density

Material	kJ/m ³ ·K	BTU/ft ³ ·°F
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Sand	1,280	19.0
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Salt	1,020	16.5
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50/50 Mix	1,155	17.94
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